



Review

Harnessing photosynthetic microorganisms for enhanced bioremediation of microplastics: A comprehensive review



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ABSTRACT

Mismanaged plastics, upon entering the environment, undergo degradation through physicochemical and/or biological processes. This process often results in the formation of microplastics (MPs), the most prevalent form of plastic debris (<1 mm). MPs pose severe threats to aquatic and terrestrial ecosystems, necessitating innovative strategies for effective remediation. Some photosynthetic microorganisms can degrade MPs but there lacks a comprehensive review. Here we examine the specific role of photoautotrophic microorganisms in water and soil environments for the biodegradation of plastics, focussing on their unique ability to grow persistently on diverse polymers under sunlight. Notably, these cells utilise light and CO₂ to produce valuable compounds such as carbohydrates, lipids, and proteins, showcasing their multifaceted environmental benefits. We address key scientific questions surrounding the utilisation of photosynthetic microorganisms for MPs and nanoplastics (NPs) bioremediation, discussing potential engineering strategies for enhanced efficacy. Our review highlights the significance of alternative biomaterials and the exploration of strains expressing enzymes, such as polyethylene terephthalate (PET) hydrolases, in conjunction with microalgal and/or cyanobacterial metabolisms. Furthermore, we delve into the promising potential of photo-biocatalytic approaches, emphasising the coupling of plastic debris degradation with sunlight exposure. The integration of microalgal-bacterial consortia is explored for biotechnological applications against MPs and NPs pollution, showcasing the synergistic effects in wastewater treatment through the absorption of nitrogen, heavy metals, phosphorus, and carbon. In conclusion, this review provides a comprehensive overview of the current state of research on the use of photoautotrophic cells for plastic bioremediation. It underscores the need for continued investigation into the engineering of these microorganisms and the development of innovative approaches to tackle the global issue of plastic pollution in aquatic and terrestrial ecosystems.

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1. Introduction

In today's society, the fundamental reliance on a wide variety of plastic materials has become indispensable in our daily lives. Their

production and use have increased exponentially over the last 40 years. However, the large manufacture of these long-chain polymers in many different and diverse applications (e.g., food packaging, agriculture, clothes, and construction) has converted them into an environmental concern due to their dispersal in all ecosystems. This quickly became a significant ecological issue, especially once waste management technologies could not follow the same pace as the plastic production consumption rate [1–4]. The dispersion mechanisms of polymers across diverse ecosystems are a complex topic that has been studied by several researchers. One study by Kane and Clare [5] comprehensively reviewed the dispersion, accumulation, and fate of microplastics (MPs) in deep-marine environments. The authors harnessed existing knowledge of seafloor microplastic distribution and integrated this with process-based sedimentological models of particle transport, providing new insights and identifying future research challenges [5]. The integration of process-based sedimentological and stratigraphic knowledge with insights from modern sedimentary systems, including biological activity within them, will provide essential constraints on MPs transfer to deep-marine environments, their distribution and fate, and the implications that these have for the wider ecosystem [5]. Prabhu et al. [6] conducted a literature survey on the presence of MPs in sediments, water, and biota samples across the globe. The survey found that MPs pervade the global seafloor, from abyssal plains to submarine canyons and deep-sea trenches.

Furthermore, the recent emergence of the COVID-19 crisis raised significant implications at multiple levels, including economic, health, and social affairs [7]. This has also caused environmental consequences connected to the increased use of safety items (e.g., protective masks, gloves, and COVID tests) based on different plastic polymers that will not be recycled soon [7–9]. Peng et al. [10] showed that since the beginning of the pandemic, an estimated 8.4 million tons of plastic waste has been generated from 193 countries. The same study also revealed that approximately 25,000 tons of plastic COVID waste, consisting of personal protective equipment (PPE) such as masks and gloves, has leaked into the ocean. Pandemic epicentres, in particular, have faced challenges in managing waste [11]. Hospital waste represents 73% of the bulk of global discharge, and 72% of the global discharge is from Asia, which indicates the need for better management of medical waste in developing countries [10].

Most of the daily objects are made from conventional plastic polymers (e.g., (i) low- and high-density polyethylene (LDPE and HDPE), (ii) polystyrene (PS), (iii) polypropylene (PP), (iv) polyethylene terephthalate (PET), and (v) polyamides (PA)), requiring several decades or centuries to be degraded in the environment [1,12]. The current procedures for global waste plastic management include (i) recycling (around 9% of globally produced plastics), (ii) incineration for energy production (12–19%), and (iii) landfilling and mismanagement (71–79%). Landfilling is the most used method because of the lower cost and the impossibility of recycling all plastics [1,12–14]. The issue of plastic pollution demands either a different approach to traditional mitigation measures or the development of new home-compostable polymers that are biodegradable and disintegrable in the environment [2,3,15,16]. The ecotoxicological implications of some 'compostable' bioplastics are not well studied in some cases or have been reported to have low environmental degradation under realistic conditions and similar toxicity to conventional plastics [15,17]. In addition, many of these may be blends or contain chemical additives, resulting in difficulty or impossibility of recycling [2,3,12]. Various countries are adopting various waste management plans, strategies, and methods [18]. However, various challenges are associated with waste management in different geographical contexts. Limited resources,

including infrastructure and funding, intersect with sociocultural complexities like public awareness, diverse practices, and communication gaps, further exacerbated by geographical and technological constraints [18,19]. These challenges can be addressed by implementing effective strategies tailored to the specific needs. Furthermore, the lack of proper waste management infrastructure and recycling facilities during the COVID-19 pandemic emergency has increased pollution with more plastic [20]. Governments and other stakeholders should work together to develop and implement sustainable practices that are environmentally friendly and economically viable. In Europe, one of the major challenges in waste management is the lack of coordination between generators, collectors, and disposal facilities [21]. In developing countries, waste management is a critical issue due to rapid population growth, inadequate funding, and lack of awareness [22].

In the past few decades, research on plastic waste pollution has focused on identifying its sources, its impact on the environment and human health, and developing effective waste management strategies [23]. However, the current challenges of plastic waste pollution are more complex and require a more comprehensive approach. The main emerging information accompanying this scientific field, especially concerning biotechnology and microbiology, has been summarised in Fig. 1.

Over the past decade, biotechnology and microbiology research has shifted significantly towards three key areas: (i) identification of plastic-specific biofilm/plastisphere [24–28]; (ii) enrichment of pathogenic bacteria, especially the members of the genus *Vibrio*, linked to a vector function of MPs [29,30]; (iii) microbial and enzymatic degradation of plastic debris from terrestrial and aquatic environments [31–37]. Once in the environment, plastic materials generally undergo degradation when exposed to physicochemical and/or biological factors (e.g., ultraviolet (UV) radiation, mechanical abrasion, temperature, and microbial degradation), as schematically summarised in Fig. 2. The environmental factors generate abrasion, shear (compressive, tensile, and flexural), and torsional impact [38–41]. Several manuscripts have reviewed the biotic and abiotic treatment of MPs [41–47]. Chen et al. [48] summarised the current technologies used to eliminate MPs from the environment, highlighting two key aspects towards this goal: (i) catalytic degradation of MPs into carbon dioxide (CO₂) and water and (ii) catalytic recycling and upcycling plastic wastes into fuels and valorised chemicals.

The abiotic methods rely on mechanical disruption, advanced oxidation, and photo-degradation [41,49,50]. Biodegradation is mainly accomplished by the activities of microorganisms, resulting in biodeterioration, depolymerization, digestion, and mineralisation [51]. Photodegradation is the most significant process that starts plastic decomposition in aquatic and terrestrial environments in direct sunlight; this method is abiotic [52]. Plastic degradation usually results in tiny particles of irregular shapes and sizes (<1 mm), defined as MPs, which are the main form of plastic debris found in the environment impacting aquatic and terrestrial ecosystems [53,54]. These can easily be disrupted and dispersed by wind and water erosion, consequently entering the food chain, which increases the potential to affect several ecosystem functions at several levels. Different factors influencing the degradation have been identified, especially in the aquatic environments [41,55,56]. It is worth noting that both abiotic and biotic processes have their advantages and disadvantages. The choice of method depends on the type of plastic waste, the environmental setting, and the desired outcome [52]. Regarding applicability, abiotic methods are more suitable for large-scale plastic waste management, while biotic methods are generally more suitable for small-scale [52,57]. Also, abiotic ones are more appropriate for degrading hard plastics [52].

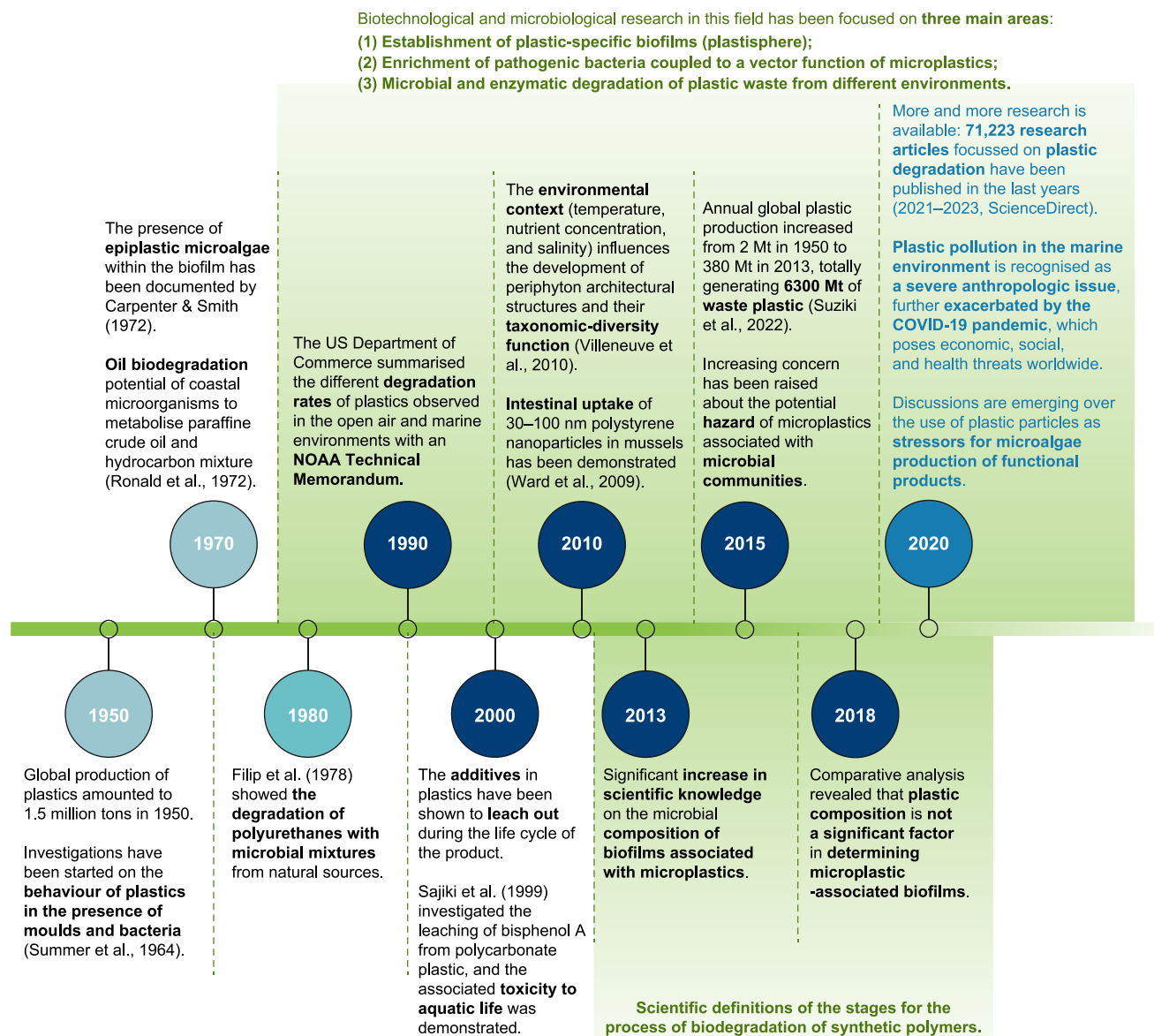


Fig. 1. Significant achievements in the scientific field of plastics and their degradation, especially with enzymes and/or microorganisms. The chronological succession of the related events in approximately 70 years is shown. Especially since 1980, an increasing number of theories and evidence regarding marine basins polluted by plastics have started to circulate. Several scientific definitions of processional stages for the biodegradation of synthetic polymers were formulated between 2008 and 2018. Studies focussing on the interactions between microplastics (MPs), nanoplastics (NPs), and microorganisms have increased in number during the last decade. NOAA: National Oceanic and Atmospheric Administration.

The environmental conditions and polymer chemistry significantly influence plastic degradation processes [12,13,53,58,59]. In principle, due to the release of toxic additives and metallic and organic compounds from their surfaces, MPs threaten the lives of several creatures: their harmful effects on marine organisms, especially the primary producers of the food chains, can influence food web consumers such as fish, aquatic birds, and even humans [60,61]. Microalgae and cyanobacteria in aquatic ecosystems could suffer from this contamination, but at the current environmental concentrations, this seems to have limited effects on parameters such as chlorophyll content, photosynthesis activity, and reactive oxygen species (ROS) [62]. The antibiotic adsorption on MPs and, therefore, the enrichment of potentially pathogenic plus antibiotic-resistant bacteria and antibiotic-resistance genes through horizontal gene transfer are other related concerns [61]. Because of all

these factors, the bioremediation of related MPs-contaminated habitats is of fundamental importance. Differently, new (bio-) degradable plastics are usually made with natural raw materials which, after the action of different living microorganisms (e.g., heterotrophic and photoautotrophic bacteria, algae, microalgae, and fungi), could be metabolised and subsequently converted into biomass and/or secondary metabolites. Hydrobiodegradable polymers, including polyesters and thermoplastic starch, undergo degradation through abiotic and biotic hydrolysis processes [63]. In contrast, polyolefins, characterised by their hydrophobic nature as hydrocarbon polymers, exhibit resistance to hydrolysis, rendering them non-hydrobiodegradable.

The various cross-linked chemicals or the disparate hydrophobicity of the plastic debris can limit the potential of living cells to both colonise and degrade the material [41]. The degradation of

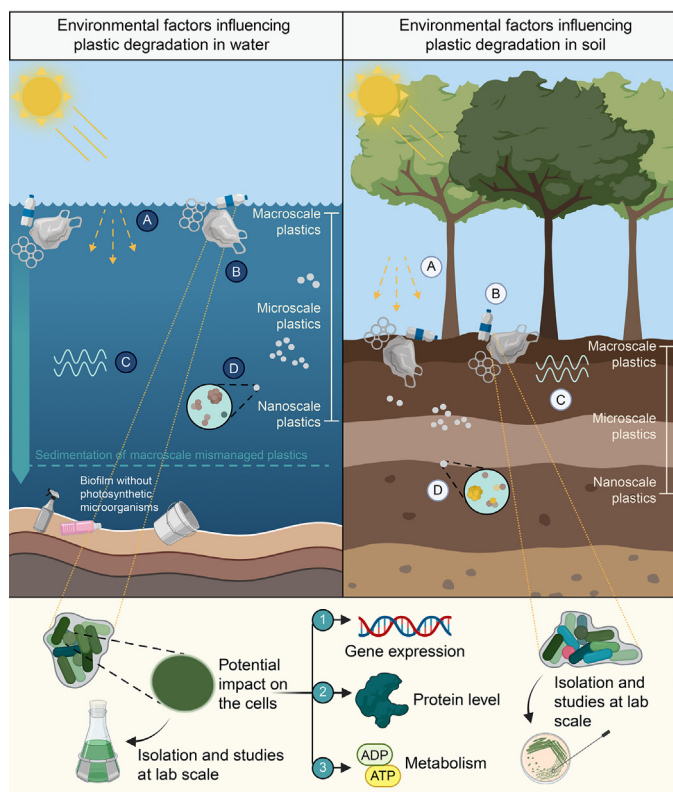


Fig. 2. Schematic overview of the biological and abiotic degradation of MPs in marine and freshwater environments (left) and in soil (right). The abiotic methods usually rely on mechanical disruption, advanced oxidation, and photo-degradation. The biofilms on the surface of different plastics are characterised by the presence of microalgae and/or cyanobacteria only if light is available. The interactions with plastics and photosynthetic microorganisms are schematised, showing the potential impacts on the cellular level for gene expression, protein level, and metabolism. A: photodegradation. B: biofilm with photosynthetic microorganisms. C: mechanical degradation. D: microbial degradation.

plastics is challenging due to their highly stable chemical structure, which makes them resistant to most types of natural degradation [64,65]. Additives (e.g., catalysts, agents for vulcanisation, curing, blowing, pigments, dyes, plasticiser, flame retardants, heat stabilisers, photostabilisers, antioxidants, and biocide) are utilised to produce the polymer for the attainment of certain physical properties of the product, such as molecular weight, molar volume, density, degree of polymerisation, the crystallinity of material, etc. [66,67]. Nonrenewable petrochemical plastics are generally highly engineered materials with precise physical properties: the properties that make plastics so versatile for humans have also caused their degradation to be challenging [64].

Plastic microbial degradation often involves using waste material as a carbon source. These metabolic processes result in an overall beneficial impact on some ecosystem's functions [12,15,17,68,69]. Such a metabolic option is an environmentally friendly and sustainable solution [9,13,58,68,70–72]. Regarding bioreactors for the bioremediation of water bodies using microorganisms and to reach some estimation for big-scale applications, the airlift bioreactor is often utilised due to its advantages over bubble column and stirred bioreactors (e.g., reduction in cell damage and greater mass transfer capacity) [73–76].

An increasing number of studies are showing wildtype [26,77–79] or engineered [80–84] strains able to degrade plastic fragments, also elucidating the related enzymatic mechanisms [85–90]. Since some of the prokaryotes and eukaryotes from the

environment could be unculturable in the laboratory, culture-independent methods are often utilised to select plastic-decomposing microorganisms: the metagenome analysis and in silico mining led to a deeper investigation towards the finding of efficient enzymes for MPs bioremediation [61]. The main wildtype and engineered microorganism able to degrade plastic debris, reported in literature, are shown in Table 1. Further, applying photosynthetic bacteria and unicellular eukaryotic algae, especially with engineered strains, may represent a possible option for plastic degradation in terrestrial and aquatic ecosystems [72,90–92]. Independently from the growth in open ponds or enclosed bioreactors, their main requirements are the availability of light, water, CO₂, and elementary nutrients. Numerous species of green algae are Generally Recognized as Safe (GRAS) for consumption by humans and animals [93]. This group offers high metabolic diversity, besides more than 8000 species, and some strains can grow in darkness with a reduced carbon source – heterotrophically [94]. Furthermore, microalgae possess the most-used type of alternative biomass in current wastewater treatment applications: these can grow and even flourish in different conditions, including a wide variety of wastewater [95,96]. The wide diversity of functional groups in their cell wall allows the binding of pollutants to the membrane surface through the biosorption phenomenon [97,98]. Other chemicals are taken up and biodegraded by living cells [96].

In this review, we summarise various photosynthetic microorganisms that have been proven capable of colonising the surfaces of different types of plastic. Using the available literature, the answers to the following scientific questions were sought: what is the specific role of photosynthetic microorganisms in water and soil for plastic biodegradation? How can they be utilised for the MPs and NPs bioremediation? How can these be engineered for this purpose? Several of these still need to be answered with further laboratory experiments, and it is not yet possible to have valid explanations. We outline various hydrolases and metabolic pathways reported in the literature, likely exploitable with engineered microalgal and/or cyanobacterial strains. Thanks to developments in the engineering of these cells in recent decades, the tools are available for expressing heterologous enzymes with hydrolysis activity against fragments of PET and other types. Through this review, the aim is also to encourage further research towards obtaining more engineered photosynthetic strains and the discovery of those natively able to potentially degrade plastic debris in polluted ecosystems. The potential of photo-biocatalytic approaches related to the degradation of plastic debris is highlighted. Furthermore, the role of microalgal-bacterial consortia for biotechnological applications against MP pollution is also included.

2. Candidates of microalgae and cyanobacteria for plastic biodegradation in terrestrial or aquatic ecosystems

Once plastic debris reaches terrestrial and aquatic ecosystems, different processes start to interact with them [64,102]. Typically, abiotic degradation pathways (UV and thermal degradation, hydrolysis, and oxidation processes) are the first to appear and are often chronologically connected with microbial degradation. The structure of the polymer influences the material's degradation, and the conventional polymers (polyethylene (PE), PP, PS, polyvinyl chloride (PVC), PET and polyurethane (PU)) can be divided according to their chemistry and degradation pathways [64,102,103]. Briefly, polymers with a carbon-carbon backbone in their main chain (PE, PP, PS, and PVC) have high molecular weight and lack functional groups susceptible to degradation (oxygen, hydroxyl, and carboxylic acid). Therefore, abiotic processes have a primary role in their degradation since they are needed to initiate the insertion of favourable functional groups. For instance, photolysis

Table 1
Exemplary wildtype and engineered micro (organisms) reported in the literature are able to degrade plastic debris.

Organism	Polymer	Experimental conditions	Outcome	Reference
Wildtype				
<i>Ideonella sakaiensis</i>	PET	Isolation from field samples	Degraded amorphous PET at ambient temperature and assimilation of its degradation monomers due to PET hydrolase (IsPETase)	[77]
<i>Moraxella</i> sp. Strain and <i>Oleispira antarctica</i>	PET	Isolation from the Antarctica environment	PET degradation at ambient temperatures (25 °C)	[99]
<i>Pseudomonas</i> and <i>Bacillus</i> consortium	PET	Isolated from petroleum-polluted soils	PET degradation	[100]
<i>Klebsiella</i> sp.	PVC	Larva's gut microbiota from <i>Spodoptera frugiperda</i>	Depolymerization and utilisation of PVC as sole energy source	[78]
<i>Enterobacter Cupriavidus</i>	PHB	Mesophilic conditions Thermophilic conditions	Methane production from PHB	[79]
<i>Moorella</i>	PHB,		Methane production from PLA	
<i>Tepidimicrobium</i>	PLA			
<i>Clostridium</i>	TPS	Mesophilic conditions	Methane production from TPS	
<i>Rhodanobacter</i> sp.	PE	Inoculation on soil suspensions	Under co-culture, the ability for polyethylene mulching film degradation	[26]
<i>Rs, Bacillus aryabhatai</i>				
<i>Pseudomonas pseudoalcaligenes</i>	PBAT	Incubation	Proteomic screening allowed the identify a new esterase, PpEst, that is involved in PBAT degradation.	[101]
Engineered				
<i>Phaeodactylum tricornutum</i>	PET	Microalgae transformation	Expression and secretion of PETase in the algal system under (mesophilic marine) growth conditions	[80]
<i>Chlamydomonas reinhardtii</i>	PET	Microalgae transformation	PET hydrolyzation. TPA, a fully degraded form of PET, was detected.	[81]
<i>Ideonella sakaiensis</i>	PET	Gene disruption system Improvement of the thermostability of leaf-branch compost cutinase	PETase and MHETase are essential enzymes for PET digestion. Improved PETase can reach up to 90% of PET degradation.	[83] [86]
<i>Pseudomonas putida</i>	PBAT, PET	Modified M9 minimal medium. Electrocompetent cells were prepared by a modified standard protocol	Plastic biodegradation assays with the best PET hydrolase expression constructs are genomically integrated into our monomer metabolism. This resulted in various degrees of plastic depolymerization. The surface display of the PET hydrolase and the secretion were successfully achieved.	[84]

PBAT: Polybutylene adipate co-terephthalate. PE: Polyethylene. PET: Polyethylene terephthalate. PHB: Polyhydroxybutyrate. PLA: Poly(lactic acid). PVC: Polyvinyl chloride. TPA: Terephthalic acid. TPS: thermoplastic starch.

and photo-oxidation lead to oxygen group insertion within the material backbone. The carbon-rich backbone is reduced to smaller and lower molecular weight fragments, which are more susceptible to biotic degradation [102].

Sometimes, different objects made of plastic have additives like antioxidants and stabilisers to inhibit or reduce polymer degradation. In other cases, pro-oxidants (such as oxygen and starch) increase their biodegradability by microbial enzymes [102]. The richness in oxygen groups characterising bio-based hydrophobic polymers is the main driver towards their degradation in natural environments and industrial composting facilities. In this respect, polymers with heteroatom in their main chains can be vulnerable to hydrolysis, photo- and biodegradation by microorganisms [64,102,104]. PS is the most resistant polymer in the environment and less susceptible to biodegradation due to its very high weight total fraction of carbon-carbon (C-C) backbone, while PU is instead the most prone to biodegradation due to their complex structure plus a relatively high abundance of O and N in the main chain [102,105]. Abiotic and biotic processes can also work in tandem, with an abiotic factor producing smaller particles which might be easily degradable by microorganisms via the mineralisation of oligomers. These processes may take a long time, several decades or centuries, even with pro-oxidants as catalysers. The existence of the bacteria on plastic surfaces is not always indicative that degradation of the specific polymer occurs. Solano et al. [106] observed that *Actinobacteria* and *Proteobacteria* were the most abundant on the surface of plastics with different degrees of biodegradability (white PE, oxo-degradable plastic, and polylactic acid-based plastic (PLA)). Even after 31 weeks in a composting test, none of the plastics showed any signs of degradation, like weight

loss or colour changes [106].

Despite these observations, some bacteria, fungi, and cyanobacteria showed the ability to degrade plastics in terrestrial ecosystems. Sangale et al. [107] and Kumari et al. [108] reviewed the capability of plastics biodegradation from bacterial and fungal isolates of agricultural soils or anthropised areas (e.g., dumping and recycling sites) [107]. Wide interest in microalgae and cyanobacteria as degraders of plastics arose from evidence of their ability in bioremediation under environmentally friendly and low-cost approaches [31,103,108,109]. Within an aqueous environment, depending upon the type of plastic fragment, the debris can either float at the surface or sink to the bottom. Either way, these fragments represent a new habitat for microbes, so much so that a diverse community of heterotrophs, autotrophs, symbionts, and predators usually start growing on their surfaces [110,111]. In most cases, floating plastic communities are dominated by microalgal cells, as these photosynthetic organisms are known to exude polymers, which allow for their attachment and rapid growth. Most other buoyant plastics tend to sink to the sediment after the accumulation of microorganisms colonising the surface, resulting in the biofouling process [112,113]. Although a good amount of data is available for communities developing on plastic fragments localised at the top of aqueous bodies (mainly the cyanobacteria *Phormidium* and *Rivularia*), only limited information exists on the composition of items with microbial communities sampled from the seafloor [110,114–116]. When the subsurface plastisphere was investigated, commonly observed taxa include Bacteroidetes (Flavobacteriaceae) and Proteobacteria (Rhodobacteraceae and Alcanivoracaceae) [110,115,116].

As widely reported, the interaction between MPs and

photosynthetic microorganisms likely affects aquatic environments at the ecosystem level. Indeed, due to the high surface area, availability of sunlight, and abundance of nitrogen and phosphorus within the aquatic media, plastic debris provides a superb growth matrix for microalgae. The optimal temperature for their growth varies depending on the species, but generally, this corresponds to values between 20 and 30 °C [117,118]. Temperatures exceeding the optimal range negatively impact the growth rate [117]. Regarding pH, these cells typically function optimally within the pH range of 6–9, facilitating their specific metabolic activities [119]. Resulting from their growth, eutrophication may detrimentally influence the aquatic ecosystems [111]. Cyanobacteria have been reported as the most prevalent organisms for plastic Interactions in agricultural fields and highly anthropised areas [109,120]. Microorganisms in a consortium can potentially increase plastic degradation, more than a single photoautotrophic or heterotrophic strain, due to metabolic cross-feeding for co-metabolic degradation [109,121]. In this respect, *Jacksonvillea* sp. ISTCYN1 was isolated from rice fields: the biofilm formation was reported in the presence of LDPE, HDPE, and PP as carbon sources, plus the ability to produce extracellular polymeric substances (EPS) and enzymes (e.g., laccase, esterase, lipase, and peroxidase) being involved in these processes [109]. These cells were grown in BG-11 medium adjusted to pH 7.4, incubated for 15 days at 25 ± 1 °C and light 12:12 h with an intensity of 3500 L × [109]. The growth conditions utilised for *Jacksonvillea* sp. ISTCYN1 are frequently employed for various other strains of cyanobacteria. This strain could be an effective option for plastic degradation under natural growth conditions. Generally, unless they are extremophiles, the temperature range to maintain alive photoautotrophic microorganisms on the surface of different types of plastic waste could assume values between 15 and 32 °C (pH 7–9 in aquatic environments). This is a wide range, which also contains suboptimal temperatures. While Mishra et al. [109] observed biofilm formation of *Jacksonvillea* sp. ISTCYN1 on plastics, the expression of the enzymes is not yet experimentally validated, even if these are predicted from its genome. MacLean et al. [122] found cyanobacteria (*Tychonema* CCAP 1459-11B and *Nostoc* sp.) to dominate plastic debris as a growth substrate. They suggested that cyanobacteria promote biocrust formation, increasing soil fertility by contributing to positive net production. In addition, these authors showed microbes growing in the cracks after UV-weathering and then suggested that microbes may be degrading their samples of plastics dumping sites in Germany. Finally, some phototrophic bacteria from the phyla *Actinobacteria* (e.g., genera *Actinomycetospira*, *Arthrobacter*, *Rhodococcus*, *Rubrobacter*, and *Cellulosimicrobium*) and *Proteobacteria* have been indicated with the ability to grow on plastic surface collected from microbial communities in field dumping sites [122].

Samples dominated by these cells contained a mix of plastic debris with a dark organic matrix, with *Nostocales* being the most abundant. Other species from *Arthrospira*, *Calothrix*, *Hydrocoleum*, *Lyngbya*, *Phormidium*, *Oscillatoria*, or *Spirulina* genera have been identified for potential plastic degradation [109,120,122,123]. This suggests that cyanobacteria may be vital in ecological succession as colonisers. Similar results highlight that microalgae can colonise, grow, and degrade PE through ligninolytic and exopolysaccharide enzymes in aquatic and terrestrial ecosystems [103,108,124]. To facilitate their attachment to the plastic surface, microalgae and cyanobacteria generally produce exopolysaccharides around and outside their cells or filaments; this could provide an appropriate environment for polymer degradation [125]. Besides, soil-water diatoms (*Navicula dicephala*, *Navicula minuta*, *Nitzschia intermedia*, *Synedra tabulate*) have been identified from plastic bags from dumping sites with the ability to grow on LDPE surface [120], although more research is needed. Additionally, when

biodegradable PLA and conventional PE film were incubated under different temperatures, these microorganisms were the most abundant on plastic surfaces [126].

When the potential of microalgae to colonise MPs of LDPE, HDPE, PET, or a mix of these was investigated at different sites, the cells successfully colonised this debris regardless of the condition examined. Sarmah and Rout indicated that some species (*Chlorella* sp., *Scenedesmus quadricauda*, or *Stigeoclonium tenue*) can grow on rubbish bags isolated from domestic dumping sites [120,123]. In studies by Sarmah and Rout [127] focused on the colonisation of *Oscillatoria* on submerged polyethylene in domestic sewage water of Silchar town in India, the temperature of domestic sewage water (pH varying from 5.8 ± 0.10 to 8.1 ± 0.23) ranged between 28 and 34 °C. Nava et al. [111] assessed the factors affecting the distribution of different photosynthetic microorganisms, identifying the cyanobacterium *Aphanocapsa incerta*. Besides this species, other frequent microalgae included *Cocconeis placentula* Ehr., *Peridiniopsis elpatiewskyi* (Ostenf.) Bourrelly, *Achnanthydium minutissimum* (Kütz.) Czarnecki, *Cocconeis pediculus* Ehr., and *Planktolymnobia limnetica* [111].

In addition to the abovementioned reports on the growth of microorganisms on the plastic surface and plastic degradation, some authors [13,128–131] indicated that PVC, LDPE or PLA weight loss due to potential degradation could also be related to degraded plasticisers by photosynthetic microorganisms since they could use these as a source of carbon and energy. This concept has been highlighted by Ru et al. [13] for PVC since materials based on this polymer have the highest proportion of plasticisers (up to 50%), and they are more susceptible to microbial attack. According to these authors, plastic weight loss is more related to plasticised PVC-degrading microorganisms than those capable of degrading PVC. Therefore, more analysis is needed to verify the potential plastic degradation since weight loss analysis, CO₂ evolution, and gas chromatography cannot accurately assess this circumstance. Wright et al. [132] performed a detailed proteogenomic and metabolomic analysis on two marines selected strains from plastic debris (*Halomonas* sp. ATBC28 and *Mycobacterium* sp. DBP42), allowing the identification of the enzymes and the pathways involved in the biodegradation of three plasticisers (dibutyl phthalate (DBP), bis(2-ethylhexyl)phthalate (DEHP), and acetyl-tributylcitrate (ATBC)): this integrated multi-OMIC study also revealed the mechanisms used for ester side-chain removal (esterases and enzymes involved in the β-oxidation pathway) as well as the molecular response to deal with toxic intermediates like phthalate. Furthermore, this study demonstrated the metabolic potential in the biofilms colonising plastics to effectively biodegrade related additives and flag the importance of microbes in reducing environmental plastic toxicity [132].

3. Hydrolases with activities against PET for photosynthetic microorganisms

Microalgae show promise in biodegradation processes, possessing the capability to produce a variety of enzymes, including hydrolases. The degradation of plastics by these cells offers a potential eco-friendly and sustainable solution to plastic contamination [133]. One of the main issues with plastics is the wide variety in use and how they are an integral part of our everyday lives. PET is a material denser than water and a major ocean pollutant; therefore, microbial remediation of PET polymers has been traditionally focused on these polymers, as highlighted in Table 1. According to a systematic review published in 2022, the most prevalent wild-type PET-degrading microorganisms were bacteria (56.3%, 36 genera), fungi (32.4%, 30 genera), microalgae (1.4%, the genus of *Spirulina* sp.), and invertebrate associated microbiota (2.8%) [134]. Among

fungi and bacteria, the most prevalent genera were *Aspergillus* sp. and *Bacillus* sp. Among the publications analysed by Benavides Fernández et al. [134], 51.8% (71 articles) applied wild microorganisms, and 32.8% (45 articles) exploited genetically modified microorganisms. Regarding the percentage of PET degradation, *Ideonella sakaiensis* still corresponds to the most efficient wild-type microorganism to degrade PET [134].

Many current techniques for dealing with these major plastic pollutants (PE and PET) require large amounts of energy. For this reason, and others outlined by the United Nation's Sustainable Development Goals (SDGs), additional, less intensive, and economical processing methods are in desperate need. An important factor to consider when looking for appropriate enzymatic methods for plastic breakdown is the conditions in which they operate; otherwise, the same energy input issues will be encountered. Hydrolases are a wide group of enzymes that catalyse bond cleavages by reaction with water. This class of enzymes catalyses biochemical reactions between different functional groups, including peptide or carbon-halide bonds, thioesters, and ureas. The action of these enzymes results in the addition of $-H$ or $-OH$ groups (from H_2O) to the plastic chains on the substrate's surface [135]. From the late 1970s, commercial lipases and even an esterase have been reported to hydrolyze polyester materials effectively: the discovery of specific PET hydrolases supported the interest in PET degradation by microorganisms. It stimulated the characterisation, enhancement, and identification of further analogous enzymes [136]. Most microbes able to secrete PET hydrolases are not ideal for industrial processes due to their complex genetic backgrounds and genetic modifications. If they are applied to industrial environments, ethical implications and regulatory frameworks would need to be addressed. Unfortunately, native microorganisms exhibit low expression rates of PET hydrolases, limiting their efficiency in plastic degradation. Therefore, enhancing expression levels is a crucial avenue to explore, particularly if microalgae are employed as a viable platform for plastic breakdown [137]. Work conducted in that area on *Escherichia coli* (*E. coli*) expression rates was limited by temperature stability, and this would be the next obstacle for microalgal species. Thus, genetic hosts such as *Clostridium thermocellum*, a thermophilic microorganism, could be used as they have proved to be a promising platform for lignocellulose processes [138]. Recently, there has been significant progress in specific PET hydrolases, such as lipases, cutinases, and esterases derived from biological sources [139]. A great milestone was the isolation of *Ideonella sakaiensis* 201-F6 from a waste PET recycling station, which is capable of degrading both PET and the reaction intermediate mono (2-hydroxyethyl) terephthalate (MHET), using a PETase and an MHETase [77]. The discovery and modification of PETase and MHETase have provided an important basis for the degradation of PET at ambient temperature conditions. Since the knowledge of the protein structure is fundamental for its engineering, shortly after, several manuscripts focussing on the structure of *Ideonella sakaiensis* PETase (IsPETase, EC 3.1.1.101) were published. This hydrolase possesses a strictly conserved active site based on the Ser-His-Asp catalytic triad, containing an optimal substrate binding site to hold four MHET moieties of PET. Usually, wildtype PETase exhibits an optimum pH range of 7–9 and stability around pH 6–10 [140]. When the purified PETase was applied to PET film, pH 9.0 was identified as optimal and 30 °C as the optimum temperature [141]. Furthermore, in contrast to cutinase, this enzyme demonstrates lower activity on *p*-nitrophenol-linked aliphatic esters and exhibits 5.5- to 120-fold higher activity towards PET [141–143]. TfCa, a versatile carboxylesterase derived from *Thermobifida fusca*, has exhibited an intriguing ability to hydrolyze intermediates involved in the degradation of PET; furthermore, the enzymatic structure of TfCa was recently determined, shedding

light on its functional characteristics [90]. Once more, artificial intelligence (AI) designing strategies have been used to optimise PET hydrolytic activity. With recent advances in AI, microalgal candidate hosts could be found for plastic processing, such as thermophilic species from the genera *Cyanidioschyzon*, *Galdieria*, and *Thermochromatium*. These have been studied for their ability to perform photosynthesis and produce biomass at elevated temperatures if required. They already possess heat-resistant enzymes and heat shock proteins that help them cope with thermal stress [144]. DuraPETase and FAST-PETase can significantly enhance PET degradation efficiency at mild-temperature environments, opening potentials for enzymatic plastic biodegradation at industrial-scale applications [145,146]. The schematic representation of the breakdown mechanism of PET is summarised in Fig. 3, where the polymer is proposed to degrade into CO_2 and water via several intermediates by using different microorganisms (including microalgae) as sources of hydrolases [147,148]. PET is converted into TPA, MHET, and bis (2-hydroxyethyl) terephthalate (BHET). Then, MHET is hydrolysed by MHETase to TPA and ethylene glycol.

TPA may then be metabolised to protocatechuic acid and 2-pyrone-4,6-dicarboxylic acid before entering the Krebs or Citric Acid Cycle. The degradation monomers can easily be recycled, although, at this point, the monomers may become pyruvate, oxalo-acetate, or eventually water and carbon dioxide molecules. The green microalgal species *Chlamydomonas reinhardtii* (*C. reinhardtii*) was modified to synthesise PETase. When comparing two strains of *Chlamydomonas* (*C. reinhardtii* CC-124 and CC-503), it was observed that CC-124 expressed more enzymes at 30 °C for up to four weeks [81]. After incubation, TPA was detected by high-performance liquid chromatography, and via microscopy, holes and dents were observed on the surface of the PET films. Interestingly, *C. reinhardtii* is generally recognized as safe (GRAS) and, therefore, suitable for biotechnological applications in natural habitats [81].

Due to the molecular structure of PET, *in vitro*, enzymatic degradation would be a logical step even if the purification plus preparation process of PET hydrolases is still expensive and time-consuming [137]. PET and similar plastics have high molecular weight and robust crystalline polymers. For these reasons, various thermal/chemical/mechanical pre-treatments of the plastics may be employed. The successful isolation of PETase from *I. sakaiensis* has opened the door for more studies into microbially-based research for plastic degradation. Further work should be conducted on other major plastics like PE, PP, or PS with or without pre-treatments. The marine diatom *Phaeodactylum tricorutum* with heterologous expression of PETase has been studied [80]. However, that strain did not grow as well as the green algae *Chlamydomonas* and *Chlorella* [149,150]. So far, *C. reinhardtii* is a potential leader microorganism for further research into PETase expression, being a freshwater microalga cultivable on a large scale. The rapid advances in cyanobacteria and prokaryotic cells for use as green cell factories for directly producing chemicals from CO_2 and solar energy make them the most interesting hosts as biocatalysts for the degradation and use of waste plastics [91,151].

It is also important to mention the industrial status regarding the biodegradation and biorecycling of plastic waste. Direct use of cells expressing enzymes capable of degrading polymers, especially when coupled with the ability to utilise energy from light via photosynthesis as in photoautotrophic microorganisms, is at a state closer to the laboratory than the industrial scale. The successful implementation of this technology based on alive microorganisms needs detailed knowledge of the polluted site of metabolism, growth, and function of native microbial population: this could be overcome with the utilisation of cell immobilisation, nano bioremediation, biosurfactant, metagenomics, chemotaxis, and

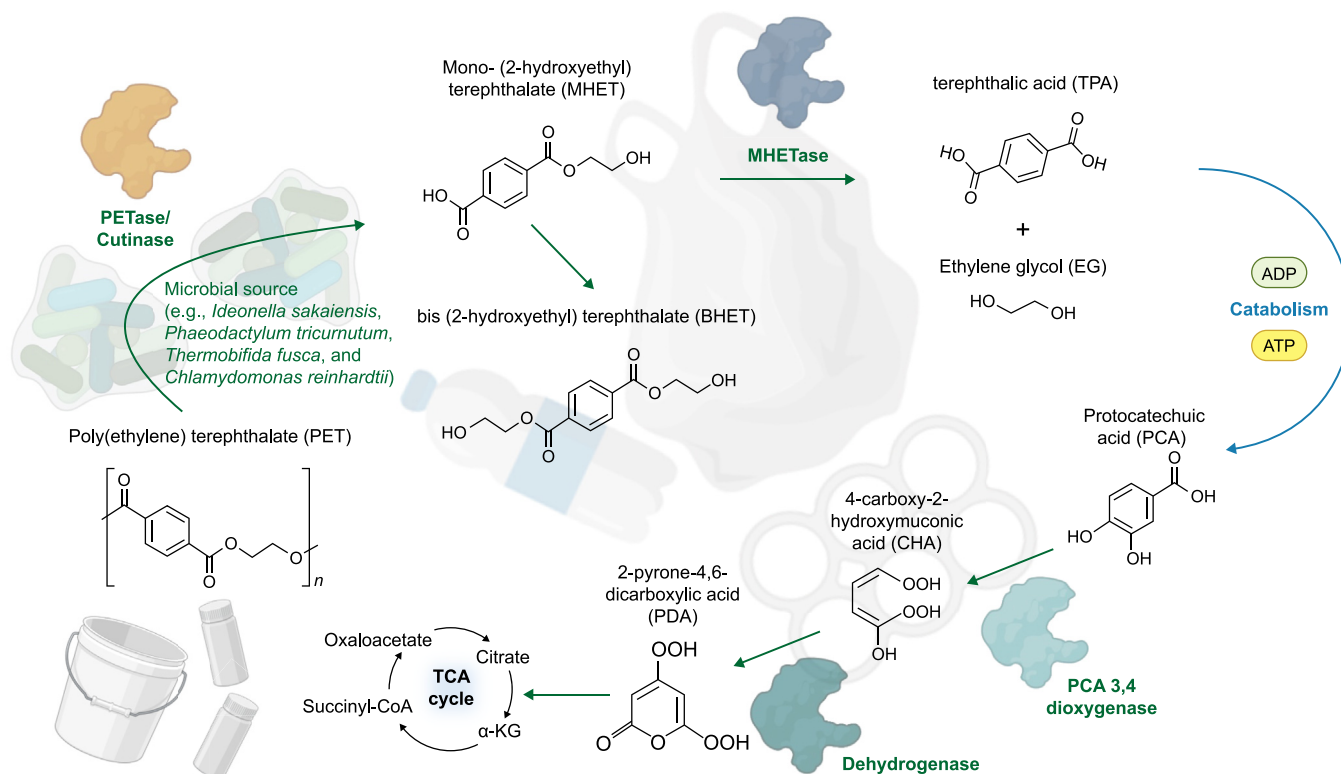


Fig. 3. Breakdown mechanism of PET using an active form of PETase derived from microalgae. Microorganisms (*Ideonella sakaiensis*, *Phaeodactylum tricornutum*, *Thermobifida fusca*, and *Chlamydomonas Reinhardtii*) were used as sources of hydrolases, which were able to catalyse the bond cleavages by reaction with H_2O . BHET: bis (2-hydroxyethyl) terephthalate. CHA: 4-carboxy-2-hydroxymuconic acid. EG: ethylene glycol. MHET: mono-(2-hydroxyethyl) terephthalate. PCA: protocatechuic acid. PDA: 2-pyrone-4,6-dicarboxylic acid. PET: poly (ethylene) terephthalate. TCA cycle: tricarboxylic acid cycle. TPA: terephthalic acid.

proteomics [152,153]. Especially regarding PET, its recycling with biocatalytic *in vitro* reactions based on engineered enzymes has notably advanced in recent years. Scaling up bioremediation enzymes for industrial use is a complex process that requires careful consideration of various factors such as the type of enzyme, the nature of the contaminant or, more specifically, the type of polymer in case of plastic waste degradation, the scale of the project, and the economic feasibility [154–156]. Considering large-scale industrial applications, the four best-performing enzymes for PET degradation were recently compared [156]. Interestingly, the next generation of LCC^{CCG} enzyme will be utilised in the world's first PET biorecycling plant commissioned in 2025 [157].

4. Photolysis, photocatalytic and photobiocatalytic transformations as tools to remediate plastic debris pollution

Pyrolysis is a conventional technology used to mitigate MPs pollution, even if this approach has shown some disadvantages: high energy consumption, generation of secondary pollution, and difficulty controlling [41,47,158]. Photolysis is another abiotic approach to degrade MPs, which is based on the presence of light. Most MPs are recorded on land or floating on the ocean surface, being exposed to sunlight for extended periods [159,160]. Different manuscripts focus on some aspects of this wide topic or MPs photolysis [161–163]. Some MPs, such as polycarbonate, are responsive to light due to the presence of photoactive groups in their structures, namely chromophores [164,165]. Alkylphenol, aromatic amines, and thiopropionate esters are frequently brought into plastic manufacturing to tolerate oxidation, which has a low effective photolysis [166,167]. Therefore, light plays a crucial role in

polymer degradation. Sunlight is known to be the natural energy source that degrades plastic waste at a very slow rate. Mimicking the role of sunlight, the photocatalytic degradation process could significantly accelerate the rate thanks to the photocatalyst that drastically facilitates the photochemical reactions involved in the degradation process [168].

Photocatalysis has long been considered for approaches regarding sustainability, such as micropollutant degradation, CO_2 reduction, and H_2 evolution reaction [169–171]. The main difference between MPs photocatalytic degradation and photolysis concerns the input of photocatalysts. For the photocatalytic degradation process, the addition of photocatalysts enriches the amount and type of ROS, stimulating depolymerization, degradation, and mineralisation in the aqueous environment [41]. Light is used as a source of energy to degrade contaminants, mostly exploiting the formation of ROS successfully [171,172]. The increasing number of published studies has highlighted the method of photocatalytic degradation as a tool to diminish MPs pollution in marine and freshwater environments [161,173]. Ouyang et al. [174] and Wang et al. [175] suggest that the use of photocatalytic technology represents a highly promising approach to transforming plastic wastes into value-added products through green and mild methods. The photocatalytic plastic waste upcycling reaction routes can be modulated by regulating experimental conditions (e.g., solvents and atmosphere), thus resulting in controllable product selectivity [175]. There are relatively few studies on photocatalytic degradation of plastic fragments. In the typical experiments, TiO_2/Pt , CdS/CdO_x and CN_x/Ni_2P photocatalysts are exploited [174]. However, TiO_2/Pt can only absorb UV light (approximately 4% of the total energy of sunlight) and are relatively

expensive, and the CdS/CdO_x quantum dots are toxic [174]. To know the details of case studies or real-world examples and investigated types of polymers, showing the practical applications of these photocatalysts, the review published by Ouyang et al. [174] contains specific information. For example, using CdS/CdO_x against 50 mg mL⁻¹ PLA, 25 mg mL⁻¹ PET, and 25 mg mL⁻¹ polyurethane (PUR), the following experimental conditions were applied: in a sealed photoreactor having the solution volume of 2 mL for 10 M sodium hydroxide (NaOH), 0.5 mM quantum dots were irradiated for 4 h at 25 °C with simulated solar light (AM 1.5G, 100 mW cm⁻²) [174,176]. Aiming for sustainability, it is necessary to design non-toxic photocatalysts driven by visible light to transform plastics into H₂ fuel and various organic chemicals. Still, the progress in the synthesis and modification methods of photocatalysts, coupled with a more in-depth understanding of photodegradation mechanisms, will enable the achievement of higher activities than those currently obtained [174,175,177].

Applying enzymes from several microorganisms and the parallel exploitation of the energy from light can correspond to a more effective degradation [178]. MPs and NPs are potentially decomposable with photocatalysis, headed for smaller molecules or additionally mineralised to CO₂. Valuable compounds may be formed due to photocatalytic CO₂ reduction [41,179], and those with smaller molecular weight could be assimilated into the anabolic pathways of different microorganisms and give rise to complete carbon cycles [180].

Photosynthesis is a biological process that utilises light energy to convert inorganic carbon into organic matter: photoautotrophic microorganisms generally have high photosynthetic and metabolic capacities and can produce a variety of valuable metabolites, such as lipids, carbohydrates, pigments, and proteins [181–183]. By exploiting the coupling with the energy obtained using available light via photosynthesis, the capability to fix CO₂, and the ability of selected enzymes to produce desired molecules, microalgae and cyanobacteria may prove to be more advantageous and sustainable as hosts for the photobiocatalytic reactions by removing the need for the external addition of co-factors. Generally, pH and temperature for these approaches have values for the optimal activity of the heterologous enzyme (e.g., pH 7–9 and 30 °C for PETase) and for the optimal function of specific intracellular metabolic pathways (e.g., pH 7.0–7.6 and generally between 20 and 30 °C for model cyanobacteria and microalgae) of the used photosynthetic microorganisms (e.g., energy in form of adenosine triphosphate (ATP) and nicotinamide adenine dinucleotide phosphate (NADPH) from photosynthesis). The methods available for their engineering allow the expression of different heterologous enzymes: the gene expressions can be tuned by either exploiting self-replicative plasmids or integrative vectors under different synthetic and native promoters of varying strength [184–188]. The ability of cyanobacteria to recycle NADPH, using light-driven water oxidation to supply electrons, makes them attractive as photobiocatalysts for different uses [189–197]. Particularly, their capacity to utilise light for the biotransformation of specific substrates into the desired products is a unique advantage compared to heterotrophic bacteria. Some studies also found the reductive ability of *in vivo* biotransformations using terpenes, hydrocortisone, chalcones, and oxophosphonates [198,199]. Studies of photobiocatalysis were also performed using microalgae like *C. reinhardtii* [200]. The developments in recombinant cyanobacterial and microalgal applications may facilitate the remediation of plastic pollution, expressing selected enzymes able to degrade these polymers and exploiting light. Especially when employing engineered strains by genetic modification or evolved adaptation, NPs might be considered as substrates for photobiocatalytic reactions: the potential application of light-driven biotransformations, converting

nanoscopic plastics particles into valuable products with photosynthetic microorganisms, could lead to an innovative scientific topic of applied biotechnology.

5. Beneficial interactions between heterotrophic bacteria and photosynthetic microorganisms in mixed-species biofilms

Most microorganisms naturally occur in mixed-species communities. This allows them to interact like mutual biofilm development [201,202]. The already mentioned plastispheres, which globally develop on plastics, rely on various interaction mechanisms involving bacteria and microalgae: this was confirmed in plastispheres of the Mediterranean, where such interactions were shown to frequently occur [203]. Sun et al. [202] recently published a detailed review focussing on biofilm for the bioremediation of MPs in contaminated freshwater environments. In the conclusion of their work, these authors mentioned that the method of degrading MPs using biofilms is feasible, even if their degradation is currently not sufficiently significant: previous studies have found that the highest degradation that can occur is approximately 20% [202]. The main reasons are (i) the structural stability of MPs, (ii) the biodegradation process of MPs is carried out in multiple steps which are not simultaneously feasible, and (iii) the fact that it is a slow process which requires a long time to be entirely completed. Future research on biofilm applied towards the degradation of MPs should consider (i) improving the speed and quality of its formation, (ii) better efficiency of its utilisation, and (iii) screening and effectively isolating different functional strains plus culturing them [202].

Future bioremediation approaches using either wild-type or recombinant microalgae and/or cyanobacteria will likely benefit from introducing beneficial bacteria, fungi or archaea in mixed-species biofilms. In this regard, various heterotrophic bacteria can substantially improve the growth and fitness of photosynthetic microorganisms' populations by providing vitamin B₁₂, phytohormones, minerals, and other important micro-compounds [204–206]. Such beneficial interactions were verified in laboratory conditions, where substantial increases in biomass production were observed when certain microalgae were co-cultivated with specific bacteria [207]. For the biotechnologically relevant algae *Scenedesmus vacuolatus* and *Haematococcus lacustris*, up to 14-fold increases in biomass production were observed [207].

Methylobacteria that elicited the growth promotion were subjected to genomic analyses and found to harbour genes involved in the synthesis of vitamins, siderophores, and plant hormones [207]. Although the response process to the widespread phytohormone indole-3-acetic acid (IAA) differs between algae and land plants, this can lead to increased growth in both [208]. Implementing these observations for plastic degradation will require targeted approaches in the next years, evaluating viable applications outside of controlled bioreactors, where inoculation can be precisely regulated to achieve and maintain specific characteristics like certain cell density. Although the exploitable interactions between fungi and microalgae have a relation to plastics bioremediation, the approaches have not yet been extensively investigated. However, recent studies have highlighted the industrial potential of their co-applications for other purposes [209–211]. Indications for species-specific co-occurrence of microalgae and fungi were also found in environmental studies [211], although most have remained unverified.

Applying microbial communities, Niu et al. [212] and Narciso-Ortiz et al. [76] showed promising achievements towards degraded plastic debris. In detail, Niu et al. [212] recently provided new insights on the natural degradation potential of MPs by microbial communities in rivers. Potential PE/PP degrading bacterial

communities were enriched and screened by outdoor and indoor culture experiments for 1150 days in this study. Furthermore, Narciso-Ortiz et al. [76] showed results from the bioprospection of a bacterial consortium composed of four bacteria (*Xanthomonas*, *Acinetobacter bouvetii*, *Shewanella*, and *Aquamicrobium lusatiense*) for hydrocarbon biodegradation in an airlift bioreactor: testing for ten days, this consortium was able to successfully degrade the maximum diesel concentration at 20 g L^{-1} . This consortium began consuming the substrate from the initial time point, accelerating its consumption before the biomass lag phase finished (0–4 days). Hexadecane was removed by the pure cultures of *Acinetobacter bouvetii* ($72 \pm 4\%$), *Xanthomonas* sp. ($46 \pm 4\%$) and *Defluviobacter. Lusatiensis* ($40 \pm 6\%$) [213]. Mixed cultures showed an enhanced hexadecane removal, particularly with the whole consortium ($79 \pm 3\%$) and the combination of *Xanthomonas* sp. And *Acinetobacter bouvetii* ($74 \pm 7\%$). The role of *A. bouvetii* coincided with that of one of the biosurfactant producers [213]. Furthermore, different strains were isolated from the soil samples, and three of the five isolated colonies could degrade PET: *Bacillus muralis*, *Brevibacterium*, and *Serratia proteamaculans*. PET debris was incubated with each strain for three days. These authors also reported the first visual evidence of PET degradation by an isolated forest-native bacterial strain, showing that *Bacillus muralis* is the most efficient degrader [76]. Changes in PET surfaces were observed by scanning electron microscopy (SEM), and the treated and control samples of PET were examined using Fourier transform infrared (FTIR) spectroscopy: a difference in transmittance percentage in the carboxylic group was observed between the samples, probably because this corresponds to the site of action for hydrocarbon-degrading enzymes [76]. No sufficient specific enzymatic or other activity values have been reported to compare it with other microorganisms capable of degrading PET. According to the systematic review by Benavides Fernández et al. [134] *Bacillus cereus* and *Bacillus gottliebii* are two species belonging to the genus *Bacillus* that can degrade PET more efficiently than other microorganisms. Still within this genus, A recent study published in 2023 reports the biodegradation of PET by *Bacillus safensis* YX8 [214]. A review published in 2021 reports the biodegradation of PET by PETase produced by genetically modified microorganisms such as *E. coli* [143]. Comparing their activities between each other, taking into account the difficulty in doing so considering that the applied conditions are different between these studies, the modified strains expressing engineered PETase by site-directed mutagenesis (often introducing modifications to the amino acid chain to enhance thermal stability, helping in maintaining activity for a longer time) show increased enzymatic activity compared to the wildtype form [142,143].

6. Conclusions

In conclusion, photosynthetic prokaryotic and eukaryotic microorganisms emerge as promising candidates for addressing plastic pollution in terrestrial and aquatic ecosystems. Their inherent advantages, including low cost, scalability, and amenability to genetic modification through mutagenesis, breeding, or transformation, position them as versatile tools for sustainable environmental remediation. Microalgae and cyanobacteria, in particular, exhibit proficiency in reducing nitrogen, phosphate, and other hazardous compounds in wastewater and serve as crucial biomanufacturing platforms for producing several biochemicals. Leveraging sunlight as their primary energy source, these microorganisms present an eco-friendly alternative to traditional systems, significantly reducing the carbon footprint associated with

the overall process.

The diverse functional groups within the cell walls of these cells enable effective biosorption of pollutants, showcasing their potential in mitigating plastic pollution. Notably, the process of binding pollutants to their surface can occur independently of their viability, while other aggregates may be actively taken up and biodegraded by living cells. Exploring microorganisms inhabiting plastic-polluted areas worldwide holds great promise for identifying strains capable of facilitating plastic interactions and promoting biodegradation. Towards future scientific studies, targeted research is imperative to unravel the full potential of genetically modified microorganisms, such as cyanobacteria and microalgae, engineered to produce enzymes like PETase and MHETase. urethanase, and alkane hydroxylase. These enzymes could serve as the foundation for transforming MPs and NPs into valuable biomass or specific bioactive compounds, thus closing the loop on plastic waste.

The use of genetically modified organisms (GMOs) in environmental applications has been debated for decades. While GMOs can potentially provide significant benefits, ethical and regulatory concerns must be considered. From an ethical standpoint, the population is generally worried about the potential risks of releasing GMOs into the environment, creating new species that could harm different ecosystems. Additionally, there is a concern that GMOs could have unintended consequences, such as creating new allergens or toxins. From a regulatory standpoint, several agencies oversee the use of GMOs in environmental applications (e.g., in the United States, the Environmental Protection Agency (EPA); in the European Union, the European Food Safety Authority (EFSA)). The direct use of genetically modified cyanobacteria and/or microalgae in natural environments contaminated by MPs and NPs does not satisfy the ethical and regulatory points.

Despite recent strides, the field of microorganism-mediated plastic remediation remains in its infancy, presenting different possibilities for expansion and exciting discoveries. While their efficacy in treating plastic pollutants in terrestrial ecosystems, marine environments, and freshwater basins is not yet fully understood, their demonstrated ability to interact as consortia or in isolation on various plastic waste surfaces underscores the need for broader scientific exploration. Regardless of the promising role of photosynthetic prokaryotic and eukaryotic microorganisms in addressing plastic pollution, some limitations and potential biases need to be considered. For instance, the effectiveness of these microorganisms in breaking down plastics depends on various factors such as the type of plastic, the environmental conditions, and the microbial community. Additionally, using these microorganisms may not be a complete solution to plastic pollution but rather a complementary approach that can be used in conjunction with other strategies. While using GMOs in environmental applications can provide significant benefits, it is important to carefully consider their ethical and regulatory implications. In this direction, the biotechnologies based on modified cyanobacteria and microalgae (e.g., *in vivo* bioremediation, photobiocatalysis, and in combination with *in vitro* approaches based on photocatalysts) should be ensured to be utilised safely and responsibly (e.g., bioremediation via bioreactors, where GMOs can degrade and assimilate plastic waste taken from contaminated environments, avoiding environmental issues due to the introduction of new microorganisms in the natural ecosystems).

In the years ahead, widening the scientific niche for the environmental treatment of plastic waste through comprehensive studies on the engineering, interactions, and potential applications of photosynthetic microorganisms promises to unlock novel

solutions to the global plastic pollution crisis.

CRedit authorship contribution statement

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Declaration of competing interest

The authors declare that they have no known competing financial interests or personal relationships that could have appeared to influence the work reported in this paper.

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